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PLAY. this substance
interferes with
hydrogen bonding in
water. surfactant.

dissolving medium
known as a solution.

solvent. a

homogeneous solution
of water and another
substance is a called
a(n)

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katelynmarcoux19. Key

Concepts: Terms in this

set (18) In soda pop,

CO_2 (g) is a _____, and

water is the _____.

solute; solvent. What

volume of 18.0 M

sulfuric acid must be

used to prepare 15.5 L

of 0.187 M H_2SO_4 ?

(Ignore ...

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Introduction to
Chemistry Section
Goals and

Introductions Section

15.1 Why Solutions

Form Goals To explain
why changes that lead
to greater dispersal of
matter are likely to
take place. To explain
why liquids tend to mix
and form solutions. To
show how you can

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predict whether
substances are soluble
in water.

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Key Section 15.1

Organic Compounds

50. Classify each of the
following as organic or
inorganic (not organic)
compounds. a. sodium

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chloride, NaCl, in table
salt inorganic b.
hexane, C₆H₁₄, in
gasoline organic c.
ethyl butanoate, CH₃C
H₂CH₂CO₂CH₂CH₃, in
a pineapple organic d.

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Kreider-Mueller. 1 of 6.
CH 1020 Exam #3
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reference see
"Chemistry: An Atoms-
focused Approach" by
Gilbert, Kirss, and
Foster. *In addition to

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reviewing this study guide, you should i) consult the Chapter Objectives posted on Blackboard and ii) PRACTICE PROBLEMS (Old exams, mock exams, end of chapter problems, etc.)

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- We can use

thermodynamicsto tell if a reaction is product- or reactant- favored.

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the..

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should be able to...

Describe Dalton's
atomic theory. Count
protons, neutrons, and
electrons in an isotope.
Know properties of
protons, neutrons, and
electrons including:

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symbol, mass number, location, and relative charge. Describe the Rutherford experiment, and how it proved the existence of the nucleus.

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vocabulary. surface
tension. surfacant.
aqueous solution.
solvent. an inward
force that tends to
minimize the surface

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area of a l... any
substance that
interferes with the
hydrogen bonding
betwee.... water that
contains dissolved
substances. the
dissolving medium in a
solution.

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Titration & pH; Chapter
16 - Reaction Energy ...

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and Solubility Equilibria
Chapter 17.

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Questions) 21 -

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Trends in Radioactivity
(10:05) 21.2 Balancing
Nuclear Reactions
(10:07)

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